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## Ab initio study of conformational properties of (*Z,E,Z*)-cyclonona-1,3,5-triene Issa Yavari<sup>a,\*</sup>, Behrooz Adib<sup>b</sup>, and K. Zare<sup>b</sup>

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*Ab initio* HF/6-31G\* and MP2/6-31G\*//HF/6-31G\* methods were used to calculate the relative energies of the important energy-minimum conformations and transition-state geometries of (*Z*,*E*,*Z*)-cyclonona-1,3,5-triene; the calculated energy barrier for ring inversion of the twist–chair ( $C_1$ ) conformation is 29.35 kJ/mol, while the barrier for swivelling of the *trans* double bond through the polymethylene bridge is 59.76 kJ/mol.

Keywords: strained molecules, Ab initio molecular orbital, cyclic trienes, conformational energy

(*Z*,*E*,*Z*)-Cyclonona-1,3,5-triene (1) have been prepared in low yield by irradiation of a pentane solution of *trans*bicyclo[4.3.0]nona-2,4-diene at 254 nm.<sup>1</sup> While much work has been dedicated to the study of conformational properties of unsaturated eight-membered rings,<sup>2–6</sup> much less effort have been devoted to unsaturated nine-membered ring systems.<sup>7–10</sup> This study was undertaken to investigate the structural optimisation and conformational interconversion pathways of the highly strained nine-membered-ring triene 1 by comparing the geometries (HF/6-31G\*) and conformational energies (MP2/6-31G\*//HF/6-31G\*). The results from MP2/6-31G\*//HF/6-31G\* calculations are used in the conformational energies discussions below.



Theoretical calculations provide energy differences and structural parameters for isolated molecules at 0 K. Therefore, theoretical calculations are not expected, in principle, to reproduce the experimental values quantitatively.<sup>11</sup> Nevertheless, it is possible to carry out *ab initio* calculations at the Hartree–Fock level, from which many properties and structures can be obtained with an accuracy that is competitive with experiment.<sup>12–15</sup> Since the theoretical results are free from intermolecular interferences, they are a valuable tool for a systematic study of conformational effects in simple organic molecules.

Six geometries were found important for a description of the conformational features of (Z, E, Z)-cyclonona-1,3,5-triene (1). Three of these geometries correspond to energy minima and three are one-dimentional energy maxima (saddle points or transition states) as shown in Fig. 1 and Table 1. The unsymmetrical twist–chair (TC,C1) conformation of 1 is predicted to be the most stable geometry (see Fig. 1 and Table 1). The energy surface for the conformational interconversion of TC with itself was investigated in detail by changing different torsional angles. The results are shown in Figure 1 and Table 1. There are three distinct, different transition states (not counting mirror images) which are required to describe the degenerate conformational dynamics in (Z, E, Z)-cyclonona-1,3,5-triene. The internal and torsional angles of these transition states are given in Table 1.

Two simple conformational processes for degenerate interconversion of twist-chair ( $\mathbf{TC}$ , $C_1$ ) conformation with itself, via the axial-symmetrical geometries, such as **TB** and boat forms, as intermediates are shown in Fig. 1. These processes are calculated to have similar energy barriers. Any of these processes when fast, introduces a time-averaged axis of symmetry.

The third conformational process, and the one with the highest barrier, is the interconversion of the **TC** conformer with its mirror image via swivelling of the *trans* double bond through the polymethylene chain. The swivelling process was simulated by rotation of the double bond about an axis containing the two adjacent double bonds. This high-energy process leads to conformational racemisation of the chiral twist–chair geometry of **1**. The calculated swivelling barrier (59.76 kJ/mol) for compound **1** is about 20 kJ/mol lower than the racemisation barrier for (*E*)-cyclononene.<sup>7</sup> Since the two conformational enantiomers of (*E*)-cyclononene are rapidly interconverted at ambient temperature, compound **1** is expected to lack chiral stability at room temperature as a result of its relatively low conformational racemisation energy.

In summary, HF/6-31G\* calculations provide a fairly clear picture of the conformational properties of (*Z*,*E*,*Z*)-cyclonona-1,3,5-triene **1** from both structural and dynamic points of view. According to these calculations, the energy barrier for ring inversion of the unsymmetrical twist-chair conformation *via* the axial-symmetrical twist-boat or boat forms is about 30 kJ/mol.The calculated barrier for swivelling of the *trans* double bond through the polymethylene chain in **TC** conformation is 59.76 kJ/mol. Thus, conformational racemization of **1** is expected to be fast at ambient temperature. It would be valuable, of course, to have direct structural data on **1** for comparison with the results of the *ab initio* calculations.

## Calculations

The *ab initio* molecular orbital calculations, were carried out using the GAUSSIAN 98 program.<sup>16</sup> Geometries for all structures were fully optimised by means of analytical energy gradients by Berny optimiser with no geometrical constraints.<sup>17</sup> The restricted Hartree–Fock calculations with the splitvalence 6-31G\* basis set which include a set of *d*-type polarisation functions on all non-hydrogen atoms were used in these calculations.<sup>18</sup> Single point energy calculations at MP2/6-31G\*//HF/6-31G\* level were used to evaluate the

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<sup>&</sup>lt;sup>†</sup> This is a Short Paper, there is therefore no corresponding material in J Chem. Research (M).

**Table 1** Calculated total and zero-point vibrational<sup>a</sup> energies (hartree), relative energy (including zero-point energy, kJ mol<sup>-1</sup>) and structural parameters for important geometries of (Z, E, Z)-cyclonona-1,3,5- triene

Feature	<b>TC</b> , <i>C</i> <sub>1</sub>	Boat, C <sub>2</sub>	<b>ΤΒ</b> , C <sub>2</sub>	( <b>TC=TB</b> ) <sup>#</sup> , C <sub>1</sub>	$(\mathbf{TC}=\mathbf{Boat})^{\#}, C_1$	Swivelling
HF/6- 31G*//HF/6- 31G*	-347.527383	-347.522397	-347.516967	-347.512932	-347.514845	-347.499459
[MP2/6- 31G*//HF/6-31G*]	[-348.858192]	[-348.855989]	[-348.849486]	[-348.847018]	[-348.846721]	[-348.835443]
ZPE	0.198914	0.199665	0.198558	0.198687	0.198960	0.200253
Erel	0.00	13.10	27.36	37.96	32.93	73.35
[Erel]	[0.00]	[5.79]	[22.87]	[29.35]	[30.13]	[59.76]
r <sub>12</sub>	1.324	1.329	1.327	1.326	1.326	1.340
r <sub>23</sub>	1.480	1.475	1.448	1.495	1.477	1.485
r <sub>34</sub>	1.320	1.323	1.319	1.320	1.322	1.324
r <sub>45</sub>	1.477	1.475	1.448	1.489	1.467	1.448
r <sub>56</sub>	1.327	1.329	1.327	1.328	1.327	1.336
r <sub>67</sub>	1.527	1.529	1.526	1.527	1.525	1.539
r <sub>78</sub>	1.561	1.551	1.548	1.549	1.579	1.571
r <sub>89</sub>	1.549	1.551	1.548	1.547	1.557	1.580
r <sub>91</sub>	1.524	1.529	1.526	1.524	1.529	1.540
$\begin{array}{l} \theta_{123} \\ \theta_{234} \\ \theta_{345} \\ \theta_{466} \\ \theta_{567} \\ \theta_{678} \\ \theta_{789} \\ \theta_{891} \\ \theta_{912} \end{array}$	118.8	118.7	115.3	114.5	116.5	127.5
	120.8	122.3	120.8	120.0	119.8	113.3
	124.2	122.3	120.8	120.8	123.6	147.7
	120.0	118.7	115.3	115.5	120.8	113.7
	130.7	127.6	124.0	124.3	131.7	128.9
	122.9	120.3	120.1	119.5	127.8	124.5
	118.3	123.2	121.0	119.3	125.7	118.2
	116.9	120.3	120.1	117.8	115.7	120.8
	126.5	127.6	124.0	122.4	125.4	131.7
$\phi_{1234}$	-54.5	-52.1	-87.1	-89.5	-59.3	10.3
$\phi_{2345}$	149.1	138.6	134.8	133.7	144.7	173.9
$\phi_{3456}$	-79.3	-52.1	-87.1	-86.1	-50.2	-177.3
$\phi_{4567}$	-0.2	7.6	-7.2	-6.8	8.6	-3.8
$\phi_{5678}$	28.4	-69.9	93.8	93.8	-40.3	30.1
$\phi_{6789}$	-82.9	60.3	-60.6	-59.1	-0.4	-83.8
$\phi_{7891}$	126.6	60.3	-60.6	-65.4	100.4	117.2
$\phi_{8912}$	-48.7	-69.9	93.8	97.8	-59.5	-69.6
$\phi_{9123}$	0.6	7.6	-7.2	-5.9	-1.5	8.1

<sup>a</sup>Zero-point vibrational energy is scaled by a factor of 0.9135 to eleminate known systematic errors in calculations.



Fig.1 Calculated MP2/6-31G\*//HF/6-31G\* strain energy (kJ/mol) profile for ring inversion of (*Z*,*E*,*Z*)-cyclonona-1,3,5-triene.

electron correlation effect in the energies and order of stability of conformers.

Vibrational frequencies were calculated at the 6-31G\* level for all minimum energies and transition states, which were confirmed to have zero and one imaginary frequency, respectively. The frequencies were scaled by a factor of 0.91 and used to compute the zero-point vibrational energies.<sup>19</sup>

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